

1 Effects of Sand Depth on Domestic Wastewater Renovation 2 in Intermittently Aerated Leachfield Mesocosms

3 José A. Amador¹; David A. Potts²; Erika L. Patenaude³; and Josef H. Görres⁴

4
5 **Abstract:** The depth of soil below the absorption trench of a septic system is considered an important factor in protection of groundwater.
6 We examined the effects of depth on the ability of intermittently aerated sand-filled leachfield mesocosms to renovate domestic waste-
7 water. Mesocosms ($n=3$) consisted of lysimeters with a headspace O_2 concentration maintained at 0.21 mol/mol and containing 7.5, 15,
8 or 30 cm of sand that were dosed with septic tank effluent every 6 h for 328 days (12 cm/d). Sand depth had no effect on pH, dissolved
9 O_2 , PO_4 , NH_4 , or BOD_5 levels in percolate water. Nitrate levels in percolate water were higher for 30 cm than for 7.5 and 15 cm during
10 the first 70 d of the experiment, after which no differences were observed. Time-averaged removal rates of N, P, fecal coliform bacteria,
11 and BOD_5 were 22–28, 13–18, 81–92, and 81–99%, respectively, and were unaffected by depth. Wastewater renovation in intermittently
12 aerated leachfield mesocosms appears to take place in a narrow zone (≤ 7.5 cm) below the infiltrative surface, with the medium below
13 contributing little to renovation.

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15 **CE Database subject headings:** Wastewater management; Septic tanks; Sand; Nutrients; Biochemical oxygen demand; Bacteria;
16 Aeration.

19 Introduction

20 Improvement of water quality in conventional on-site wastewater
21 treatment systems (OWTSs) takes place both in the septic tank,
22 where removal of solids, oil and grease, and some anaerobic di-
23 gestion takes place, and in the absorption field, or leachfield,
24 where percolation of septic tank effluent (STE) through the soil
25 allows for microbial, organic, and inorganic constituents in efflu-
26 ent to interact with soil processes. For example, net removal of P
27 in the leachfield can take place via abiotic processes, through
28 sorption and binding of PO_4 to oxides of aluminum and iron in
29 acidic soils, and through formation of insoluble precipitates in
30 alkaline soils (Beal et al. 2005; Robertson 2003; Zanini et al.
31 1998). Net nitrogen removal, when it occurs, results from the
32 combined activities of bacteria that mineralize organic N to am-
33 monium, oxidize it to nitrate, and reduce it to N_2 and N_2O , gases
34 that are readily lost to the atmosphere (USEPA 2002). Pathogenic
35 organisms are thought to be removed from STE by a combination
36 of processes that include physical filtration, adsorption to soil

particles, predation by soil fauna, and competition with resident 37
microflora (Hagedorn et al. 1981; Reddy et al. 1981). 38

The relationship between depth and particulate and dissolved 39
pollutant removal in leachfield soil is not particularly well under- 40
stood or quantified. The extent of renovation of STE in leachfield 41
soil is thought to be proportional to soil depth (Jenssen and Sie- 42
grist 1990). For example, in a pilot-scale study of intermittent 43
sand filters with depths of 30, 45, and 60 cm, Peebles et al. (1991) 44
found that levels of BOD_5 decreased significantly as depth in- 45
creased. In contrast, levels of NH_4 were not significantly different 46
at 30 and 45 cm, but were significantly lower at 60 cm (Peebles et 47
al. 1991). Stevic et al. (1999) found a significant reduction in the 48
number of *E. coli* from STE with medium depth, with 99% of the 49
bacteria removed in the top 12 cm of 80 cm long columns packed 50
with round light weight aggregate, with higher removal observed 51
at greater depths. 52

In the case of chemical and biological processes—that are con- 53
trolled by interactions with the surfaces of soil particles, micro- 54
organisms, and/or enzymes—greater depth is thought to translate 55
into a longer residence time within soil pores, increasing the prob- 56
ability of an occurrence of the appropriate process. Filtration 57
theory predicts that retention of particulates in porous media— 58
such as soil—is a first-order process, with the concentration of 59
particulates decreasing exponentially with depth (Acostas and 60
Castillejos 2000). Thus, passage of STE through a greater depth 61
of soil increases the probability that pathogenic organisms en- 62
counter pores with dimensions smaller than their own diameter, or 63
that they pass through sufficiently narrow pore necks whose sur- 64
faces they may diffuse to and subsequently adsorb on. These as- 65
sumptions form part of the rationale behind restrictions on the 66
separation distance between the infiltrative surface of the absorp- 67
tion trench and the seasonal high groundwater table by local and 68
regional regulatory agencies in the United States and elsewhere. 69

In a previous, laboratory-scale study, we found that intermit- 70
tent aeration of leachfield mesocosms filled with 30 cm of quartz 71

¹Professor, Dept. of Natural Resources Science, Univ. of Rhode Is-
land, Kingston, RI 02879 (corresponding author). E-mail:
jamador@uri.edu

²President, Geomatrix, LLC, 385 Roast Meat Hill Rd., Killingworth,
CT 02419. E-mail: epotts@geomatrixllc.com

³Research Assistant, Dept. of Natural Resources Science, Univ. of
Rhode Island, Kingston, RI 02879. E-mail: erikanic@etal.uri.edu

⁴Research Associate Professor, Dept. of Natural Resources Science,
Univ. of Rhode Island, Kingston, RI 02879. E-mail: josefgorres@
hotmail.com

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Table 1. Mean ($n=10$) Values of Water Quality Parameters and Loading Rates of Septic Tank Effluent (STE) Inputs into Leachfield Mesocosms

Parameter	Mean value (s.d.)	Loading rate
Temperature (°C)	18.1 (1.6)	NA
pH	6.8 (0.2)	NA
Dissolved O ₂ (mg/L)	0	NA
BOD ₅ (mg/L)	230 (48)	27.6 g/m ² /d
Total N (mg N/L)	47.5 (6.3)	5.7 g N/m ² /d
NO ₃ (mg N/L)	0.0	0
NH ₄ (mg N/L)	37.1 (9.3)	4.5 g N/m ² /d
Total P (mg P/L)	7.9 (1.9)	0.9 g P/m ² /d
PO ₄ (mg P/L)	3.9 (1.2)	0.5 g P/m ² /d
SO ₄ (mg S/L)	6.9 (3.7)	0.8 g S/m ² /d
Fecal coliforms (CFU/100 mL)	5.89×10^5	7.07×10^8 CFU/m ² /d

72 sand improved the removal of N, BOD₅, and fecal coliform bac-
73 teria relative to un-aerated soil (Potts et al. 2004). These improve-
74 ments were observed in the absence of a conventional restrictive
75 layer—or biomat—and for N removal were apparent only at high
76 hydraulic loads (12 cm/d) (Potts et al. 2004). The effects of aera-
77 tion were presumed to be due to removal of electron acceptor
78 limitations by the periodic introduction of O₂ following applica-
79 tion of anaerobic wastewater to the infiltrative surface. This
80 would allow for the establishment of ammonium oxidation to
81 nitrate—a necessary step prior to denitrification—and support a
82 larger community of microbivorous fauna thought to be involved
83 in pathogen removal, as well as more efficient and complete uti-
84 lization of BOD₅. The structure and function of the biotic com-
85 munity of these soils differed markedly, with a larger and more
86 diverse microbial and faunal community found in intermittently
87 aerated soil (Amador et al. 2006). Of particular interest with re-
88 spect to STE renovation was the presence of larger numbers of
89 microorganisms and microbivorous fauna in soil at 0–4 cm than
90 at 4–13 cm in intermittently aerated soil, which indicates that
91 biological activity is concentrated in a fairly narrow band below
92 the infiltrative surface. Together, these results suggest that inter-
93 mittent aeration may reduce the depth requirements for successful
94 renovation of STE in leachfield soil.
95 The objective of the present study was to assess the extent to
96 which water quality improvements in intermittently aerated leach-
97 field soil depend on soil depth.

98 Materials and Methods

99 Facility

100 The study was conducted at a domestic wastewater research fa-
101 cility in southeastern Connecticut, described previously (Potts et
102 al., 2004). Briefly, it consists of a laboratory adjacent to a two-
103 story, two-family home built in 1983. The home was fitted with a
104 conventional septic system that was installed in 1996. The septic
105 tank had a maximum capacity of 4,733 L and was not pumped
106 during the course of the study. The home was inhabited contin-
107 uously by three to six people. A summary of the nutrient loading
108 rates and measured chemical, microbiological, and physical char-
109 acteristics of septic tank effluent during the course of the experi-
110 ment is found in Table 1.

111 A schematic diagram of the experimental setup is shown in
112 Fig. 1. All of the effluent from the septic tank was diverted to a
113 pump station and stored in an air-tight high-density polyethylene

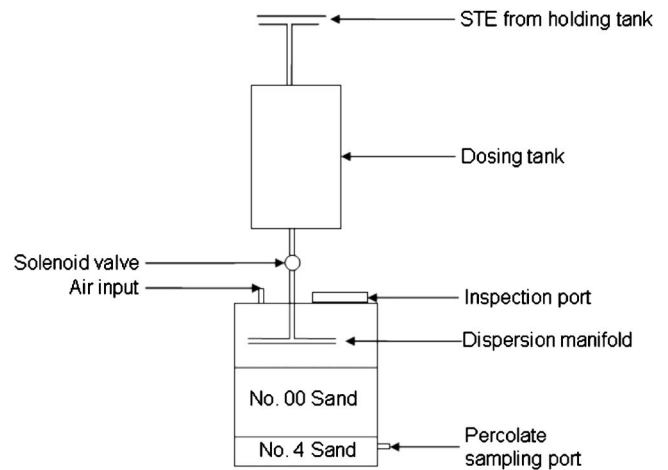


Fig. 1. Schematic diagram of experimental mesocosms used to study effects of sand depth on STE renovation

(HDPE) tank (1,325 L maximum capacity) housed in a climate-
114 controlled (17 to 19°C) room above the laboratory. The contents
115 of the tank were mixed every 2 h using a pump. STE from the
116 tank was pumped through a PVC manifold to a series of dosing
117 tanks in the laboratory. Cylindrical HDPE dosing tanks (30.5 cm
118 inner diam, 45.7 cm height) had a maximum capacity of 38 L and
119 were dosed every 6 h. Dosing was regulated using solenoid
120 valves. Dosing tank overflow was allowed to drain completely
121 until the desired dose volume was retained. STE from the dosing
122 tanks flowed by gravity into a series of mesocosms (described
123 below).
124

Depth Treatments

125 We determined the effects of sand depth with mesocosms built
126 using cylindrical HDPE tanks (43.2 cm inner diam, 45.7 cm
127 height) with fittings for air and water inputs, sampling of perco-
128 late water, and an inspection port (Fig. 1). STE was delivered to
129 the surface of the sand through a perforated horizontal, 1.91 cm
130 diam PVC manifold to attenuate the impact of delivery. STE was
131 applied to the mesocosms at a rate of 12 cm/d for the duration of
132 the experiment.
133

Treatments consisted of 7.5, 15, or 30 cm of silica sand (No.
134 00; diam 0.71–0.21 mm; uniformity coefficient <1.6; U.S. Silica
135 Co., Berkeley Springs, WV) placed on top of 7.5 cm of No. 4
136 silica sand (diam 4.75–1.40 mm; uniformity coefficient <1.8).
137 The volume above the infiltrative surface constituted headspace.
138 Each treatment was replicated three times. A blower was used
139 to introduce ambient air at regular intervals into the headspace of all
140 mesocosms to maintain an O₂ level of 0.20–0.21 mol/mol, which
141 resulted in a pressure of ~2.5–6.7 kPa. The experiment was run
142 continuously Sept. 11, 2003 to Aug. 4, 2004 (328 d). Percolate
143 water samples were taken 14, 42, 70, 98, 132, 160, 188, 216, 244,
144 272, 300, and 328 days after the start of the experiment.
145

Sampling, Processing, and Analysis

146 Water sampling and processing procedures are described in Potts
147 et al. (2004). Briefly, STE samples were collected from a valve in
148 the input stream (purged by allowing 1–2 L of STE to flow
149 through) and placed in autoclaved plastic bottles. Samples of per-
150 colate water from the mesocosms were collected in 3 L Tedlar
151 (■, ■) bags (2 mil thick, SKC, Inc., Eighty Four, Pa.). The bag
152

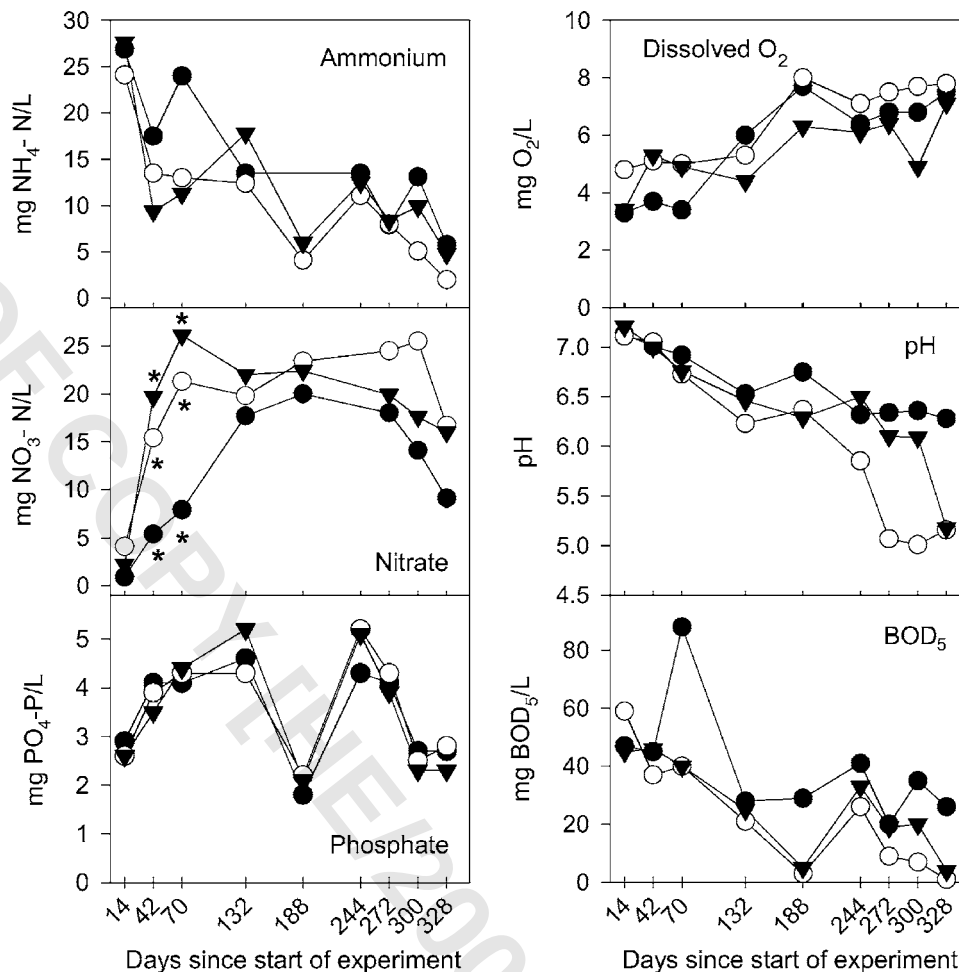


Fig. 2. Percolate water pH and concentration of BOD₅, dissolved O₂, NH₄, NO₃, and PO₄ as a function of time in intermittently aerated mesocosms with different sand depths. Values are means (*n*=3). (*) indicates values were significantly different.

153 was connected to the mesocosm outlet using Tygon tubing. To
 154 ensure that water samples were exposed to an atmosphere with
 155 the same composition as that found in the mesocosms, a connec-
 156 tion was made from the headspace of each mesocosm to the
 157 drainage line connected to the sampling bag. Sampling of water
 158 from the mesocosms started coincident with a dosing event.

AQ: 159 All water samples were analyzed immediately for dissolved
 #3 160 oxygen (DO) using the azide modification of the Winkler titration
 161 method (APHA 1998), and for the presence of Fe²⁺ using EM
 162 Quant (■, ■) Iron (Fe²⁺) test strips (EM Industries, Inc., Gibb-
 163 stown, N.J.), with the remaining volume kept at 4 °C during trans-
 164 port to the laboratory in Kingston, RI (~1 h). Unfiltered water
 165 samples were used to determine 5-day biochemical oxygen de-
 166 mand (BOD₅) using manometric respirometry [OxiTop (■, ■)
 167 BOD system; WTW, Fort Myers, FL], pH (model UB-10 pH
 168 meter; Denver Instruments, Denver, CO), fecal coliforms and *E.*
 169 *coli* [membrane filtration procedure; APHA (1998)], and total N
 170 (TN) and total P (TP) content using the persulfate digestion
 171 method (APHA 1998). The remaining sample was filtered (GF/F,
 172 25 mm diam; Whatman Intl. Ltd., Maidstone, England) and ana-
 173 lyzed for NH₄⁺, NO₃⁻, PO₄³⁻ using an automated nutrient analyzer
 174 (Flow Solution IV, Alpkem, College Station, Tex.) and for SO₄²⁻
 175 using the barium chloride turbidimetric method (APHA 1998).
 176 STE samples were analyzed in triplicate on every sampling
 177 date, whereas a single percolate water sample was analyzed per
 178 replicate mesocosm per sampling date.

Statistical Analyses

179

A one-way analysis of variance (ANOVA) of water quality pa- 180
 rameters for STE as a function of time indicated significant dif- 181
 ferences on days 98 and 216 of the experiment. As such, values 182
 for these dates were excluded from our analysis. Differences 183
 among depth treatments were determined at the 95% confidence 184
 level using a one-way ANOVA on ranks and Tukey’s test for 185
 means separation (*P*<0.05). 186

Results

187

Ammonium concentration in percolate water declined over the 188
 course of the experiment from 25–27 mg N/L to less than 189
 5 mg N/L, with no significant differences observed among depths 190
 on any sampling date (Fig. 2). The concentration of nitrate in 191
 percolate water increased initially at all depths, with nitrate levels 192
 that were significantly different among sand depths after opera- 193
 tion for 42 and 70 days (Fig. 2). No significant differences in NO₃ 194
 levels were observed among depths on any subsequent dates. Medi- 195
 um depth had no significant effect on the pH of percolate water, 196
 which declined steadily during the experiment (Fig. 2). Depth did 197
 not have a significant effect on pH on any sampling date. Initial 198
 values of DO in percolate water ranged from 3.7 to 4.7 mg/L, 199
 and DO values increased during the course of the experiment, 200

201 with final values ranging from 7.3 to 7.5 mg/L (Fig. 2). No sig-
 202 nificant differences in DO were observed among treatments on
 203 any sampling date. Levels of BOD₅ in percolate water decreased
 204 in all treatments during the course of the experiment, and no
 205 significant differences were observed among depth treatments on
 206 any sampling date (Fig. 2). The concentration of PO₄ in percolate
 207 water exhibited a great deal of variation during the course of the
 208 experiment; however, there were no significant differences among
 209 treatments on any sampling date (Fig. 2). We did not observe
 210 treatment differences in the concentration of SO₄ in percolate
 211 water among treatments and no Fe²⁺ was detected in percolate
 212 water from any of the treatments (data not shown).
 213 Removal rates for N varied during the experiment and were
 214 not significantly different among treatments on any sampling date
 215 (Fig. 3). When averaged over the course of the experiment, N
 216 removal rates (s.d.) were 28.0% (8.7), 21.6% (8.0), and 22.5%
 217 (8.0) for sand at depths of 7.5, 15, and 30 cm, respectively. Sand
 218 depth had no statistically significant effect on the of rate removal
 219 of P on any sampling date (Fig. 3). As was the case for N re-
 220 moval, rates of removal for P varied throughout the course of the
 221 experiment. Values of P removal averaged over the course of the
 222 experiment were 17.5 (8.0), 13.3 (6.8), and 15.5 (6.3) for 7.5, 15
 223 and 30 cm, respectively. The rate of removal for BOD₅ increased
 224 over the course of the experiment in all treatments, with no sig-
 225 nificant differences observed among depths (Fig. 3). Removal
 226 rates for BOD₅ averaged over the course of the experiment were
 227 81.4% (9.2), 90.6% (9.3), and 89.0% (7.0) for 7.5, 15, and 30 cm,
 228 respectively. Fecal coliform removal rates also varied throughout
 229 the course of the experiment, and no significant differences were
 230 observed among treatments on any sampling date (Fig. 3). Aver-
 231 aged over the course of the experiment, fecal coliform mean re-
 232 moval rates for 7.5, 15, and 30 cm were 82.4% (17.7), 92.3%
 233 (5.6), and 85.7% (9.5), respectively.

234 Discussion

235 The time-averaged performance of the intermittently aerated
 236 leachfield mesocosms was comparable to that observed by others
 237 in terms of removal of N, BOD₅, and fecal coliform bacteria in
 238 tests using leachfield porous materials with an equal or greater
 239 medium depth than in the present study (Duncan et al. 1994;
 240 Harrison et al. 2000; Magdoff et al. 1974; Pell and Nyberg
 241 1989a,b; Rodgers et al. 2004; van Cuyk et al. 2001). Sand depth
 242 did not appear to have a significant effect on water quality param-
 243 eters or contaminant removal rates, regardless of whether differ-
 244 ences were evaluated on individual sampling dates or over the
 245 course of the experiment. Furthermore, temporal patterns of per-
 246 colate water quality parameters and removal rates were similar
 247 regardless of depth. These results run counter to conventional
 248 wisdom with respect to wastewater renovation in porous media,
 249 which equates greater depth with increased contaminant removal.
 250 The insensitivity of removal processes for total N and BOD₅ to
 251 medium depth—which involve the activities of microor-
 252 ganisms—may result from the establishment of conditions that
 253 support the necessary microbial processes in a narrow (≤ 7.5 cm)
 254 area below the infiltrative surface. Foremost among these is the
 255 recurring increase in the level of O₂ at the infiltrative surface
 256 brought about by intermittent aeration. In the case of BOD₅, the
 257 microbial oxidation of organic carbon is more energetically effi-
 258 cient when microorganisms employ O₂ as the terminal electron
 259 acceptor (Fuhrmann 2005). Analysis of the structure of the micro-
 260 bial community of intermittently aerated and unaerated sand me-

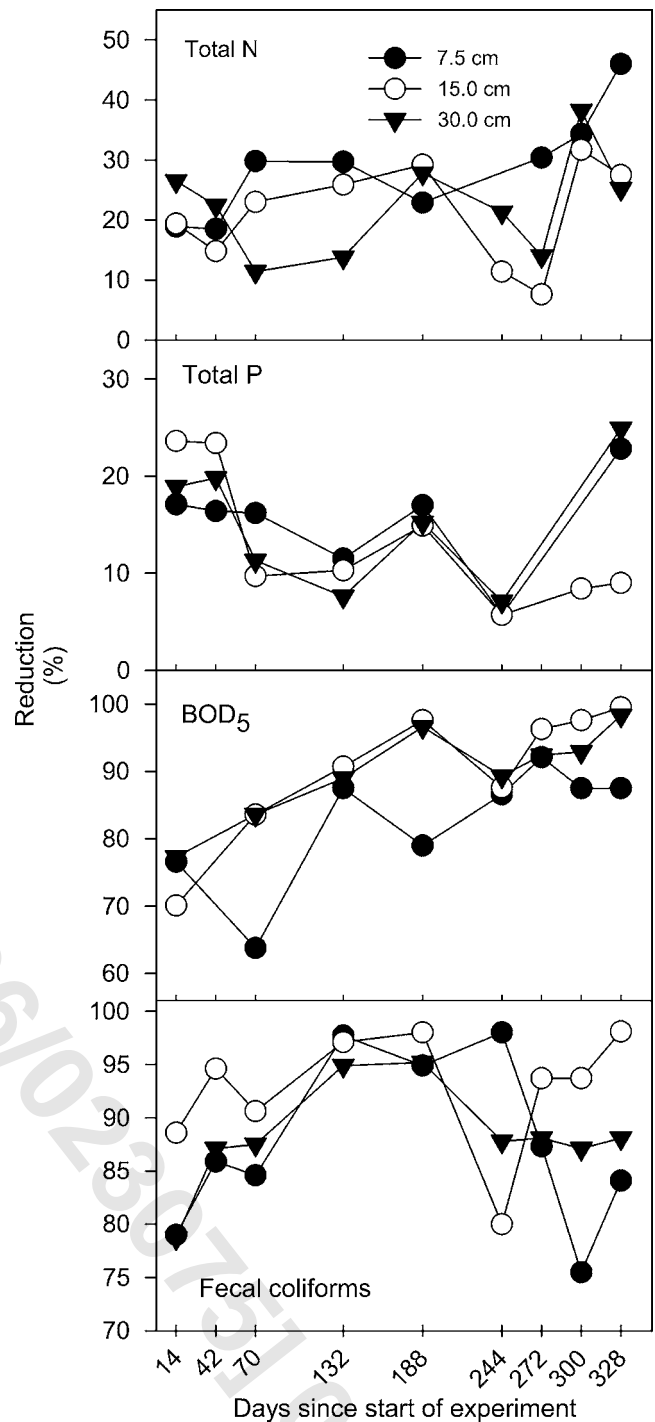


Fig. 3. Removal of fecal coliform bacteria, BOD₅, total P, and total N as a function of time in intermittently aerated mesocosms with different sand depths. Values are means ($n=3$).

socosms in an experiment similar to the present study (Amador et al. 2006) found a larger active microbial biomass at 0–4 cm than
 at 4–13 cm in intermittently aerated soil, suggesting that micro-
 bacterial activities are more concentrated in the area immediately
 below the infiltrative surface. The combination of periodic in-
 creases in availability of O₂ and a larger active microbial popu-
 lation near the infiltrative surface may explain the insensitivity of
 BOD₅ removal to soil depth.

In contrast to our results, Peeples et al. (1991) found that re-
 moval rates for BOD₅ in sand filters receiving domestic wastewa-

271 ter increased with depth, with mean values of 53, 69, and 75%
 272 reported for depths 30, 45, and 60 cm, respectively. BOD₅ re-
 273 moval rates in the present experiment were consistently higher
 274 (Fig. 3) than those reported by Peeples et al. (1991) even at the
 275 lowest depth (7.5), suggesting that intermittent aeration may be
 276 able to compensate for depth of medium for BOD₅ removal.

277 Initially total N removal likely involves the buildup of micro-
 278 bial biomass N. However, the intermittent introduction of air also
 279 supports microbial oxidation of NH₄ to NO₃, which initially was
 280 found to increase with sand depth. This increase may be due to
 281 longer fluid retention times in deeper mesocosms, which would
 282 expose nitrifying populations to NH₄ for longer periods of time,
 283 allowing for faster growth. If we assume the microbial population
 284 will reach steady state with respect to N removal for growth, net
 285 N removal then requires reduction to N₂O and N₂ by denitrifiers.
 286 Insensitivity of N removal to soil depth may stem from the estab-
 287 lishment of conditions that promote nitrification and denitrifica-
 288 tion in the zone immediately below the infiltrative surface at
 289 different times. In this conceptual model, nitrification takes place
 290 in the porous medium during periods of aeration that alternate
 291 with periods of anoxia caused by the introduction of wastewater
 292 containing high concentrations of organic carbon substrates into
 293 the same zone where NO₃ is produced. Thus, assuming that deni-
 294 trifying bacteria are present in the wastewater and/or the sand, all
 295 of the conditions necessary for N removal by denitrification can
 296 be met over a short distance. This mechanism is similar to N
 297 removal in wastewater treatment plants using a sequencing batch
 298 reactor (Henze et al. 1997), with the nitrification and denitrifica-
 299 tion steps separated in time.

300 The main abiotic mechanism of phosphorus removal in leach-
 301 field soil is thought to involve sorption and binding of PO₄ with
 302 iron and aluminum oxides and oxyhydroxides on the surface of
 303 soil particles (Robertson 2003; Zanini et al. 1998). However, as
 304 indicated in Potts et al. (2004), the quartz sand employed in our
 305 experiments does not contain appreciable amounts of these ox-
 306 ides, making this an unlikely removal mechanism. Straining of
 307 particulate P and microbial growth both likely contribute to P
 308 removal in our systems. In addition, our results may be explained
 309 by the involvement of enhanced biological phosphate removal
 310 (EBPR) processes observed in activated sludge. Removal of P in
 311 the activated sludge process via EBPR requires alternating aero-
 312 bic and anaerobic conditions (Mino et al. 1998). Net removal of P
 313 in association with EBPR can result from precipitation with Ca²⁺
 314 present in wastewater as long as the pH remains near neutral
 315 (Maurer et al. 1999). In the absence of other abiotic removal
 316 mechanisms, P removed by EBPR in our sand mesocosms has the
 317 potential to migrate below this zone as microorganisms die and
 318 biomass P is mineralized. However, the mineralogy of particles in
 319 leachfield soils is generally such that abiotic reactions with oxides
 320 of aluminum and iron will result (Robertson 2003; Zanini et al.
 321 1998), reducing the potential for migration.

322 If EBPR is the principal removal mechanism, the lack of P
 323 removal in response to sand depth may be explained by the fact
 324 that soil immediately below the infiltrative surface can be ex-
 325 pected to have the highest concentrations of O₂ during aeration
 326 periods as well as the longest periods of anoxia following inputs
 327 of wastewater and prior to aeration. Reports of simultaneous re-
 328 moval of N and P from wastewater via denitrification and EBPR
 329 in sequencing batch reactors (Gieseke et al. 2002; Zeng et al.
 330 2003) lend support to the notion that both of these processes may
 331 have been active in our mesocosms.

332 Straining is believed to be involved in the removal of fecal
 333 coliform bacteria in leachfield soils (Reddy et al. 1981). If this is

the case, the absence of a significant effect of soil depth on fecal
 coliform removal indicates that straining takes place over a rela-
 tively short distance. In addition, aeration may promote condi-
 tions that are adverse to survival of fecal coliform bacteria in soil
 below the infiltrative surface. For example, in a previous experi-
 ment, we found that soil in the top 4 cm of intermittently aerated
 lysimeters had significantly larger numbers of bacterivorous
 nematodes and protozoa—believed to be involved in removal of
 fecal coliform bacteria—than soil at 4–13 cm (Amador et al.
 2006). Development of a biomat that restricts water movement is
 known to improve fecal coliform removal, especially in coarse
 media (Stevic et al. 2004). However, the absence of a conven-
 tional biomat in intermittently aerated leachfield mesocosms
 (Potts et al. 2004) did not hinder their ability to remove fecal
 coliforms, with removal rates >99% observed. Differences be-
 tween rates reported by Potts et al. (2004) and the more modest
 removal rates (82 to 91%) observed in the present study may be
 attributed to differences in the duration of the experiments. The
 systems in Potts et al. (2004) were run for 24 months versus 11
 months in the present study. Longer running times may allow for
 development of biotic and abiotic conditions that further enhance
 bacterial removal.

Our results suggest that intermittent aeration of leachfield soil
 results in renovation of wastewater over a short distance below
 the infiltrative surface. This has important implications for design
 and permitting of innovative OWTS technologies. The proposed
 renovation mechanisms for N and P, involving alternating aerobic
 and anaerobic conditions, may circumvent the limitations im-
 posed by the regulatory requirement of arbitrary distances be-
 tween the ground water and the infiltrative surface without
 affecting the level of wastewater renovation. Intermittently aer-
 ated leachfield technology may be of use in soils deemed unsuit-
 able for installation of conventional OWTS because of a shallow
 water table. Implementation of this technology will likely require
 reevaluation of land use regulations governing site suitability for
 treatment and disposal of domestic wastewater in decentralized
 systems, as has been suggested for other innovative OWTS, such
 as shallow trench low-pressure pipe systems (Winkler and Feiden
 2001).

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